

EAST Search History

Ref #	Hits	Search Query	DBs	Default Operator	Plurals	Time Stamp
L1	55	(562/824).CCLS.	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	OFF	2006/06/06 07:23
L2	263	fluorine adj liquid	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L3	5786	fluorine adj gas	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 07:23
L4	5999	(fluorine adj liquid) or (fluorine adj gas)	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L5	1191	elemental near3 fluorine	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L6	6896	((fluorine adj liquid) or (fluorine adj gas)) or (elemental near3 fluorine)	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 06:32
L7	0	US-06586626-\$.DID.	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L8	173	560/184	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L9	3	(fluorine adj liquid) and 560/184	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L10	1	US-0658662-\$.DID.	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L11	1	"0056694".pn.	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41

EAST Search History

L12	7	((562/825).CCLS.) and ((fluorine adj liquid) or (fluorine adj gas))	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L13	5	((562/825).CCLS.) and (elemental near3 fluorine)	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L14	17	(((fluorine adj liquid) or (fluorine adj gas)) or (elemental near3 fluorine) and (US-6586626-\$.DID. OR US-3900372-\$.DID. OR US-4524032-\$.DID. OR US-4868318-\$.DID. OR US-4996369-\$.DID. OR US-5093432-\$.DID. OR US-5322903-\$.DID. OR US-5466877-\$.DID. OR US-0488142-\$.DID. OR US-5571870-\$.DID. OR US-5578278-\$.DID. OR US-5674949-\$.DID. OR US-5753776-\$.DID. OR US-6093860-\$.DID. OR US-6255536-\$.DID.)	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L15	29	US-6586626-\$.DID. OR US-3900372-\$.DID. OR US-4524032-\$.DID. OR US-4868318-\$.DID. OR US-4996369-\$.DID. OR US-5093432-\$.DID. OR US-5322903-\$.DID. OR US-5466877-\$.DID. OR US-5488142-\$.DID. OR US-5571870-\$.DID. OR US-5578278-\$.DID. OR US-5674949-\$.DID. OR US-5753776-\$.DID. OR US-6093860-\$.DID. OR US-6255536-\$.DID.	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L16	2	"6586626".pn.	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L17	2	("3900372").PN.	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	OFF	2006/06/06 05:41
L18	137	(562/825).CCLS.	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	OFF	2006/06/06 05:41
L19	934	fluorosulfonyl	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41

EAST Search History

L20	1041	fluorosulfonyl\$	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L21	1251	L19 or L20	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L22	4271	(sulfonylfluoride) or (sulfonyl adj fluoride)	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L23	58	L21 same L22	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L24	4	L4 and L23	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L25	29	US-6586626-\$.DID. OR US-3900372-\$.DID. OR US-4524032-\$.DID. OR US-4868318-\$.DID. OR US-4996369-\$.DID. OR US-5093432-\$.DID. OR US-5322903-\$.DID. OR US-5466877-\$.DID. OR US-0488142-\$.DID. OR US-5571870-\$.DID. OR US-5578278-\$.DID. OR US-5674949-\$.DID. OR US-5753776-\$.DID. OR US-6093860-\$.DID. OR US-6255536-\$.DID.	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L26	2	"6255536".pn.	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 05:41
L27	692	((fluorine adj liquid) or (fluorine adj gas)) or (elemental near3 fluorine) and ((sulfonylfluoride) or (sulfonyl adj fluoride)).clm.	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 06:34
L28	225	((fluorine adj liquid) or (fluorine adj gas)) or (elemental near3 fluorine) and ((sulfonylfluoride) or (sulfonyl adj fluoride)).clm.	US-PGPUB	OR	ON	2006/06/06 06:34
L29	225	((fluorine adj liquid) or (fluorine adj gas)) or (elemental near3 fluorine) and ((sulfonylfluoride) or (sulfonyl adj fluoride)) and decompos\$.clm.	US-PGPUB	OR	ON	2006/06/06 06:35
L30	4	(fluorine and ((sulfonylfluoride) or (sulfonyl adj fluoride)) and decompos\$).clm.	US-PGPUB	OR	ON	2006/06/06 06:35

EAST Search History

L31	137	(562/825).CCLS.	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	OR	OFF	2006/06/06 07:23
L32	242677	fluorine	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 07:23
L33	57	I31 and I32	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2006/06/06 07:24

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FILE LAST UPDATED: 5 Jun 2006 (20060605/ED)

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=> sulfonylfluoride or (sulfonyl fluoride)
90 SULFONYLFLUORIDE
6 SULFONYLFLUORIDES
96 SULFONYLFLUORIDE
(SULFONYLFLUORIDE OR SULFONYLFLUORIDES)
29947 SULFONYL
18 SULFONYLS
29958 SULFONYL
(SULFONYL OR SULFONYLS)
249387 FLUORIDE
44668 FLUORIDES
265073 FLUORIDE
(FLUORIDE OR FLUORIDES)
981 SULFONYL FLUORIDE
(SULFONYL (W) FLUORIDE)
L1 1064 SULFONYLFLUORIDE OR (SULFONYL FLUORIDE)

=> fluorine
100846 FLUORINE
533 FLUORINES
L2 101142 FLUORINE
(FLUORINE OR FLUORINES)

=> l1 and l2
L3 83 L1 AND L2

=> 73-83 ti
158804 73
162706 83
344835 TI
1144 TIS
345678 TI
(TI OR TIS)

L4 0 73-83 TI
(73 (W) 83 (W) TI)

=> d 17 73-83 ti

L7 NOT FOUND

The L-number entered has not been defined in this session, or it has been deleted. To see the L-numbers currently defined in this session, enter DISPLAY HISTORY at an arrow prompt (=>).

=> d 13 73-83 ti

L3 ANSWER 73 OF 83 CAPLUS COPYRIGHT 2006 ACS on STN
TI Photochemical etching of silicon dioxide surface layer

L3 ANSWER 74 OF 83 CAPLUS COPYRIGHT 2006 ACS on STN
TI β -Sultones containing fluorine. XXXII.
Fluorosulfo-containing triazines

L3 ANSWER 75 OF 83 CAPLUS COPYRIGHT 2006 ACS on STN
TI β -Sultones containing fluorine. XXVII.
 α -Hydrotetrafluoroethanesulfonic acid derivatives

L3 ANSWER 76 OF 83 CAPLUS COPYRIGHT 2006 ACS on STN
TI Aromatic fluoro derivatives. XXIV. Fluorine-substituted benzenesulfonyl fluorides

L3 ANSWER 77 OF 83 CAPLUS COPYRIGHT 2006 ACS on STN
TI Fluorine-containing carbon compounds

L3 ANSWER 78 OF 83 CAPLUS COPYRIGHT 2006 ACS on STN
TI Structure and reactivity of fluorine-containing β -sultones

L3 ANSWER 79 OF 83 CAPLUS COPYRIGHT 2006 ACS on STN
TI Reactive dyes containing fluorosulfonyl groups

L3 ANSWER 80 OF 83 CAPLUS COPYRIGHT 2006 ACS on STN
TI Fluorine-containing β -sultones. II. Hydrolysis of tetrafluoroethane β -sultone

L3 ANSWER 81 OF 83 CAPLUS COPYRIGHT 2006 ACS on STN
TI Toxic fluorine compounds. VIII. ω -Fluoroalkanesulfonyl chlorides and fluorides

L3 ANSWER 82 OF 83 CAPLUS COPYRIGHT 2006 ACS on STN
TI Substituted pyridines

L3 ANSWER 83 OF 83 CAPLUS COPYRIGHT 2006 ACS on STN
TI Sulfonyl fluorides of amino azo dyes

=> l1(1012

MISSING OPERATOR 'L1(L0L2'

The search profile that was entered contains terms or nested terms that are not separated by a logical operator.

=> l1(1)12

L5 25 L1(L) L2

=> d 15 10-25 ti

L5 ANSWER 10 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Dissolution characteristics of acidic groups for 157-nm resist

L5 ANSWER 11 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Process for the preparation of fluorinated derivatives of nucleosides or

sugars

L5 ANSWER 12 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Chemical modification of a Nafion sulfonyl fluoride precursor via in situ
sol-gel reactions

L5 ANSWER 13 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Chemical treatment of glass surface with fluorine-containing silanes

L5 ANSWER 14 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Photopolymerized Epoxide Copolymer Thin Films with Surfaces Highly
Enriched with Sulfonyl Fluoride Groups

L5 ANSWER 15 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI New polyfluoroalkoxysulfonyl fluorides. Part VIII. Alcoholic and polymeric
derivatives

L5 ANSWER 16 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Separation of perfluoroalkyl sulfonyl fluoride

L5 ANSWER 17 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI A useful synthesis of ω -iodoperfluoroalkanesulfonyl fluorides and
perfluoroalkane- α , ω -bis-sulfonyl fluorides

L5 ANSWER 18 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Improved acid electrolytes - the synthesis and structure of
fluorine-containing sulfonic acids for fuel cells. Final report July
1987-August 1988

L5 ANSWER 19 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Process for manufacture of hypofluorites and bishypofluorites

L5 ANSWER 20 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI New sulfonyl fluoride esters

L5 ANSWER 21 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Synthesis and evaluation of fluorine-19-labeled sulfonyl
fluorides as probes of protease structure: α -chymotrypsin

L5 ANSWER 22 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Fluorine-containing β -sultones. 50. Geminal bis(fluorosulfonyl)-
containing compounds

L5 ANSWER 23 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Inorganic volatile fluorides obtained from electrical decomposition of
sulfur hexafluoride in a quartz tube

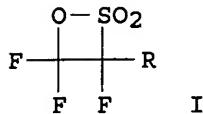
L5 ANSWER 24 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Fluorine-containing β -sulfones. 46. 2-
Hydrohexafluoropropane-2-sulfonyl fluoride

L5 ANSWER 25 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Substituted pyridines

=> d 15 20 ti fbib abs

L5 ANSWER 20 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI New sulfonyl fluoride esters
AN 1987:477290 CAPLUS
DN 107:77290
TI New sulfonyl fluoride esters
AU Khalilolahi, Jilla; Mohtasham, Javid; Lerchen, Megan E.; Sheets, Roger M.;
Gard, Gary L.
CS Dep. Chem., Portland State Univ., Portland, OR, 97207, USA

SO Inorganic Chemistry (1987), 26(14), 2307-9
CODEN: INOCAJ; ISSN: 0020-1669
DT Journal
LA English
OS CASREACT 107:77290
GI



AB New **sulfonyl fluoride esters** $\text{FSO}_2\text{CF}_2\text{C}(\text{O})\text{ORf}$ [$\text{Rf} = \text{CF}_3\text{CH}_2, (\text{CF}_3)_3\text{C, C}_6\text{F}_5$], $\text{FSO}_2\text{CF}(\text{CF}_3)\text{C}(\text{O})\text{ORf}$ [$\text{Rf} = \text{CF}_3\text{CH}_2, (\text{CF}_3)_2\text{CH, C}_6\text{F}_5$], and $\text{FSO}_2\text{CF}(\text{CF}_3)\text{C}(\text{O})\text{OR}$ ($\text{R} = \text{Et, allyl}$) were prepared by using the perfluoro sultones I ($\text{R} = \text{F, CF}_3$) with appropriate polyfluoro/perfluoro alcs. or hydrocarbon alcs. in the presence of NaF . In a like manner the diester $[\text{FSO}_2\text{CF}(\text{CF}_3)\text{C}(\text{O})\text{OCH}_2]^2$ was prepared from ethylene glycol and I ($\text{R} = \text{CF}_3$). Unlike the fluorinated esters with **fluorine** at the alkoxy α -carbon atoms, these esters are stable in the presence of fluoride ion at 25° or higher temperature. Their IR, NMR, and mass spectra are reported.

=> d 15 1-9 ti

L5 ANSWER 1 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Simple transformation of thymine 1-[3-hydroxy-2-(phosphonomethoxy)propyl] derivatives to their 1-[3-fluoro-2-(phosphonomethoxy)propyl] counterparts

L5 ANSWER 2 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Pyrimidine 1-[2-(phosphonomethoxy)propyl] derivatives: their syntheses and utilization as potent inhibitors of thymidine phosphorylase (PD-ECGF) from Sd-lymphoma

L5 ANSWER 3 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Method for producing **fluorine**-containing unsaturated **sulfonyl fluoride**

L5 ANSWER 4 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Preparation of stabilized fluoropolymer

L5 ANSWER 5 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Process for production of fluorinated sulfonyl fluorides useful for ion exchange resins and chemical-resistant substances

L5 ANSWER 6 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Process for preparation of perfluorinated sulfonyl fluorides

L5 ANSWER 7 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Method for producing **fluorine**-containing **sulfonyl fluoride** compound

L5 ANSWER 8 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Process for preparing (per)fluorohalogen ethers by the reaction of acyl fluorides with halogenated 1,2-difluoroethylenes

L5 ANSWER 9 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
TI Fluoro type surfactants

=> d 15 3,5-7 ti fbib abs

L5 ANSWER 3 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
 TI Method for producing fluorine-containing unsaturated
 sulfonyl fluoride
 AN 2005:729629 CAPLUS
 DN 143:193723
 TI Method for producing fluorine-containing unsaturated
 sulfonyl fluoride
 IN Sugiyama, Akinari; Ichihara, Kazuyoshi; Shinoki, Noriyuki; Mantani,
 Toshiya; Kondou, Masahiro
 PA Daikin Industries, Ltd., Japan
 SO PCT Int. Appl., 20 pp.
 CODEN: PIXXD2
 DT Patent
 LA Japanese
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2005073182	A1	20050811	WO 2005-JP1005	20050126
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				

JP 2004-25768

A 20040202

OS CASREACT 143:193723; MARPAT 143:193723
 AB A method for producing a fluorine-containing unsatd.
 sulfonyl fluoride represented by the chemical formula
 RfSO_2F (wherein Rf is a fluorine-containing hydrocarbon group having
 at least one unsatd. bond and may contain at least one element selected
 from oxygen, nitrogen and sulfur) is characterized in that a
 fluorine-containing unsatd. sulfonyl chloride represented by the chemical
 formula RfSO_2Cl (wherein Rf is as defined above) is reacted with at least
 one fluorinating agent selected from alkylamine hydrofluoride, pyridine
 hydrofluoride, and polyvinylpyridine hydrofluoride. By this method, a
 fluorine-containing sulfonyl fluoride having an
 unsatd. bond can be com. advantageously produced at low cost.
 Furthermore, this method enables to produce the fluorine-containing
 unsatd. sulfonyl fluoride in a simple procedure with
 high selectivity and high yield. Thus, 20.0 g $\text{CF}_2:\text{CFOCF}_2\text{CF}_2\text{SO}_2\text{Cl}$ was
 added dropwise at 1.67 g/min to 33.5 g $\text{Et}_3\text{N} \cdot (\text{HF})_3$ with stirring at
 22° during which the liquid temperature rose from 22° to 33°.
 After completion of the addition, the resulting mixture was stirred for
 .apprx.1 h to give a reaction mixture with three phases $\text{Et}_3\text{N} \cdot (\text{HF})_n$ ($n = 4-6$)
 (liquid phase)/ $\text{Et}_3\text{N} \cdot \text{HCl}$ (solid phase)/ $\text{CF}_2:\text{CFOCF}_2\text{CF}_2\text{SO}_2\text{Cl}$ (product liquid phase)
 (bottom phase) which was distilled by a simple distillation to give 96.0%
 $\text{CF}_2:\text{CFOCF}_2\text{CF}_2\text{SO}_2\text{Cl}$.

RE.CNT 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L5 ANSWER 5 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
 TI Process for production of fluorinated sulfonyl fluorides useful for ion
 exchange resins and chemical-resistant substances
 AN 2005:29302 CAPLUS
 DN 142:114654
 TI Process for production of fluorinated sulfonyl fluorides useful for ion
 exchange resins and chemical-resistant substances
 IN Murata, Koichi; Okazoe, Takashi; Murotani, Eisuke
 PA Asahi Glass Company, Limited, Japan

SO PCT Int. Appl., 48 pp.

CODEN: PIXXD2

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2005003082	A1	20050113	WO 2004-JP9769	20040702
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
	RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
EP	1642890	A1	20060405	JP 2003-271071	A 20030704
	R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, FI, RO, CY, TR, BG, CZ, EE, HU, PL, SK			EP 2004-747237	20040702
				JP 2003-271071	A 20030704
				WO 2004-JP9769	W 20040702
US	2006111584	A1	20060525	US 2005-318978	20051228
				JP 2003-271071	A 20030704
				WO 2004-JP9769	A1 20040702

AB Title process comprises (i) oxidizing a compound YSRAERB with an oxidizing agent containing a halogen atom as the essential constituent into a compound XSO₂RAERB, (ii) converting the obtained compound into FSO₂RAFEFRBF by reacting with fluoride when X is a fluorine atom or after conversion of X into a fluorine atom when X is a halogen atom other than fluorine in a liquid phase, and (iii) decomposing this compound into a compound FSO₂RAFCOF, wherein RA = a divalent organic group such as alkylene; RB, RBF = a monovalent organic group such as perfluoroalkyl; E = CH₂OCO; Y = a monovalent organic group such as cyano; X = a halogen atom; RAF = a divalent organic group obtained by fluorinating RA; and EF = CF₂OCO. Thus, 21.7 g 3-bromo-1-propanol and 64.1 g 2,3,3,3-tetrafluoro-2-[1,1,2,3,3,3-hexafluoro-2-(heptafluoropropoxy)propoxy]-propanoyl fluoride reacted, the resulting compound was reacted with thiocyanic acid potassium salt, reacted with chlorine, substituted with fluoride, perfluorinated, and decomposed to give perfluoro(3-fluorosulfonyl)propionyl fluoride, which was reacted with hexafluoropropene oxide in the presence of cesium fluoride and diglyme, potassium hydrogen carbonate and glyme were added therein, and heated at 180-210° to give FSO₂(CF₂)₃OCF:CF₂.

RE.CNT 9 THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L5 ANSWER 6 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN

TI Process for preparation of perfluorinated sulfonyl fluorides

AN 2005:29287 CAPLUS

DN 142:113432

TI Process for preparation of perfluorinated sulfonyl fluorides

IN Murata, Koichi; Okazoe, Takashi; Murotani, Eisuke

PA Asahi Glass Company, Limited, Japan

SO PCT Int. Appl., 52 pp.

CODEN: PIXXD2

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2005003062	A2	20050113	WO 2004-JP9779	20040702
	WO 2005003062	A3	20050324		

W:	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW		
RW:	BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG		
EP 1640362	A2 20060329	JP 2003-270412	A 20030702
R:	AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, FI, RO, CY, TR, BG, CZ, EE, HU, PL, SK	EP 2004-747247	20040702
US 2006106252	A1 20060518	JP 2003-270412	A 20030702
		WO 2004-JP9779	W 20040702
		US 2006-322396	20060103
		JP 2003-270412	A 20030702
		WO 2004-JP9779	A1 20040702

OS MARPAT 142:113432
 AB This invention pertains to a method for producing fluorinated sulfonyl fluorides with general formula of $(FSO_2)_nRAF(EF_1)_m$ [wherein RAF = a (fluorinated) $(n+m)$ valent organic group having two or more carbon atoms; EF₁ = one valent organic group; $n \geq 2$; $m \geq 1$] via fluorination and decomposition. For example, $(FSO_2CF_2)_2CFCOF$ was prepared in a multi-step synthesis starting from $(BrCH_2)_2CHCO_2H$ in good yield. This invention provides a convenient method to prepare perfluorinated sulfonyl fluorides at low cost with industrial advantages.

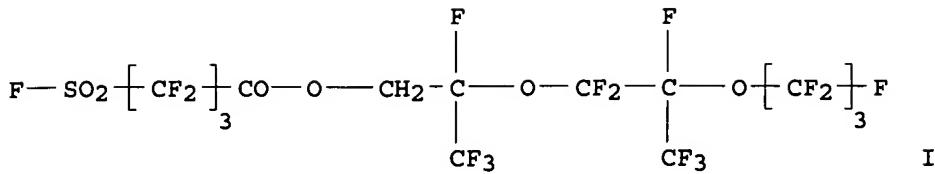
L5 ANSWER 7 OF 25 CAPLUS COPYRIGHT 2006 ACS on STN
 TI Method for producing fluorine-containing sulfonyl fluoride compound
 AN 2004:927161 CAPLUS
 DN 141:395193
 TI Method for producing fluorine-containing sulfonyl fluoride compound
 IN Murata, Koichi; Okazoe, Takashi; Murotani, Eisuke
 PA Asahi Glass Company, Limited, Japan
 SO PCT Int. Appl., 34 pp.
 CODEN: PIXXD2

DT Patent
 LA Japanese
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2004094365	A1	20041104	WO 2004-JP5874	20040423
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
	RW: BW, GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				

OS CASREACT 141:395193
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JP 2003-119874 A 20030424



AB A method for producing a fluorine-containing sulfonyl fluoride compound which comprises oxidizing RB-E-RA-S-S-RA-E-RB to form XSO₂-RA-E-RB, reacting the oxidation product with fluorine in a liquid phase to form FSO₂-RAF-EF-RBF, and decomposing the fluorination product to prepare FSO₂-RAF-COF [RA = divalent organic group; RAF = divalent organic group or the like; RB = monovalent organic group; RBF = monovalent organic group or the like; E = -COOCH₂-; EF = -COOCF₂- ; X = halo] is disclosed. The method is almost free from major conventional difficulties associated with the production of the above sulfonyl fluoride, and also allows the production of fluorine-containing sulfonyl fluoride compds. having various mol. structures and being useful as a raw material for an ion exchange resin or the like with good efficiency at a low cost. For example, a mixture of compound I (4.2 g), e.g., prepared from (S(CH₂)₃COOH)₂ in 4 steps, and NaF (0.03 g) was stirred at 140 °C for 10 h and analyzed by GC-MS. The yield of FSO₂(CF₂)₃COF was 73.7%. Of note, disclosed compds. are useful intermediates for ionic exchange resin.

RE.CNT 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

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